CYCLOADDITION OF DICHLOROKETENE WITH FUNCTIONALIZED CYCLOALKENES. SYNTHESIS OF BICYCLO [4.2.0] OCTANONE-3-YL DERIVATIVES AND OF 3.4-DICARBOMETHOXY-1-METHYLBICYCLO [4.2.0] OCTAN-7-ONE.

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Abstract: Dichloroketene reacts with some cycloalkenes in which an ester function is in homoallylic or in an even more remote position with respect to sp carbon atoms. In fact acetate and benzoate of 3,5,5-trimethylcyclohex-3-en-1-ol and 1,2-dicarbo= methoxy-4-methyl-cyclohex-4-ene undergo 2+2 cycloaddition of dichloroketene to produce the corresponding bicyclo [4.2.0] octan= one derivatives in 60-65% yield. In the latter case the process occurs regiospecifically to give 3,4-dicarbomethoxy-1-methylbi= cyclo [4.2.0] octan-7-one as product after dechlorination. The Hassner zinc dehalogenation method of generating dichloroketene is the best procedure.

The $\left[2+2\right]$ cycloaddition of dichloro= ketene with cyclopentadienes and olefins is widely recognized as a remarkably powerful method for construction of four membered rings. The α, α -dichlorocyclobu= tanone derivatives can be readily dechlorinated with zinc or subjected to other useful transformations.

Although this reaction has been the subject of intensive investigations, to date it has only seen limited applications with functionalized cycloalkenes.

The effectiveness of dichloroketene cy= cloadditions togheter with our interest in exploring new techniques for non pho= tochemical synthesis of bicyclo 4.2.0 al= kane derivatives, provided incentive for an investigation of cyclization process with cycloalkenes bearing activating functionality. We report here the results of a study concerning reactions of dichlo=roketene with cycloalkenes in which a sup= plementary function is in homoallylic, or in an even more remote position with

respect to sp^2 carbon atoms. We chose compounds <u>la-c</u>, <u>2</u> and <u>3</u> as substrates for this study.

RESULTS

Dichloroketene, prepared in situ by deha=
logenation of trichloroacetylchloride
with activated zinc in presence of pho=
sphorus oxychloride, reacted with som=
pound <u>la</u> in refluxing diethyl ether.
Work-up of mixture after 10-12 h provi=
ded 8,8-dichloro-1,5,5-trimethylbicyclo=
[4.2.0]octan-7-one-3-yl acetate (<u>4</u>) and
7,7-dichloro-1,5,5-trimethylbicyclo [4.2.0]
octan-8-one-3-yl acetate (<u>5</u>) in 63% yield
(Scheme 1). Bicyclo derivatives <u>4</u> and <u>5</u>

proved to be rather unstable even when kept in refrigerator under nitrogen. They were dechlorinated with excess zinc dust in saturated methanol solution of ammo= nium chloride. After 6 h of reflux it was possible to obtain quantitative amounts of 1,5,5-trimethylbicyclo [4.2.0]octan-7-one-3-yl acetate (6) and the isomeric bi= cyclo [4.2.0]octan-8-one-3-yl acetate (7). Direct treatment of the reaction mixture

between <u>la</u> and dichloroketene without separation of compound <u>4</u> and <u>5</u> produced compounds <u>6</u> and <u>7</u> in the same yield and without modification of the isomers ratio (ratio 6:7 = 1.78).

In the same conditions the benzoyl ester <u>lb</u> underwent [2+2] cycloaddition of dichloroketene, and successively deha=logenation to give compounds <u>lO</u> and <u>ll</u> in 61% yield (ratio <u>lO:11</u> = 1.85), (Sche=me 2).

Similar treatment of 1,2-dicarbometho=
xy-4-methylcyclohex-4-en (3) produced on=
ly 3,4-dicarbomethoxy-8,8-dichloro-1-me=
thylbicyclo [4.2.0]octan-7-one (12) in 65%
yield. Compound 12 underwent dehalogena=
tion when treated with zinc dust in reflu=
xing saturated solution of ammonium chlori=
de and 3,4-dicarbomethoxy-1-methylbicyclo
[4.2.0]octan-7-one (13) was separated in all
most quantitative yield, (Scheme 3).

All compounds showed the expected spec= tral data (IR, ¹H NMR) and analytical properties. In addition Bayer-Williger oxidation with metachloroperbenzoic acid (MCPBA) was performed on regioisomer <u>11</u> to have additional evidence for struc= tural assignment. The bicyclolactone <u>14</u> was obtained in 90% yield.

It is noteworthy that dichloroketene, prepared in situ by dehydrohalogenation of dichloroacetylchloride with triethyl= amine, reacted with <u>la</u> and <u>lb</u> in reflu= xing n-hexane to give complex mixtures. The IR spectra of these reaction mixtu= res showed a strong C=O absorption at 1805 cm⁻¹ (cyclobutanone system) while the stretching absorption at 1730 cm⁻¹ for the ester function was absent.

Reactions performed with tetrahydrofus ranyl derivative <u>lc</u> and on methyltetras hydrophtalic anhydride <u>2</u> were unsuccessful. Complex reaction mixtures were obstained and all our efforts to separate some significative product were unsuccessful.

We expected that our findings may en= courage a much greater use of dichloro= ketene in non photochemical synthesis of bicyclo n.2.0 alkane derivatives.

EXPERIMENTAL

Proton NMR spectra were recorded at 60 MHz on a Varian EM360 L instrument and at 100 MHz on a Varian XL-100 operating in the CW mode. ¹H NMR shifts are given in parts per milion from Me₄Si in CDCl₃ solvent. IR spectra were recorded with a Perkin-Elmer 257 spectrophotometer. Mi=croanalyses were performed by using C,H, N Analyzer Model 185 from Hewlett-Packard Co.. 3,5,5-Trimethylcyclohex-3-en-1-ol (1) was prepared in high yield by lithium aluminium hydride reduction of β-phorone, easily available by the method of Marx. 1,2-Dicarbomethoxy-4-methylcyclohex-4-en

(2) was prepared by esterification of the 1:1 adduct 3 of Diels Alder cycload=dition of isoprene and maleic anhydride? The activation of zinc dust was performed according a procedure previously reported. Analytical grade solvents (Carelo Erba)) were used without further purification.

3,5,5-Trimethylcyclohex-3-en-1-ol($\underline{1}$). A 500 ml three necked flask equipped with a condenser, additional funnel; magnetic stirrer, and No inlet, was charged with 2.73 g (7.2x10 mol) of LiAlH, and dry diethyl ether (170 ml). The suspension was stirred under N, and a solution of 15 g $(10.8 \times 10^{-2} \text{ mol})$ of β -phorone in dry diethyl ether (50 ml) was added dropwi= se during 1 h. When addition of solution was complete, the mixture was refluxed for half an hour. The cooled reaction mi= xture was treated with water (Caution!) and , successively, with a cold aqueous 5% HCl solution, washed with brine and dried(Na₂SO₄). The solvent was removed at reduced pressure to leave an oil we= ighing 15.2 g. Vacuum distillation affor= ded 14.6 g of product (96%): bp 80-82°C (6.0 mm_{Hg}); IR(neat) 3340(s),1670(m)cm⁻¹; 1H NMR & 5.1(s,1H),4.0(m,1H),2.2-1.5(m, 4H),1.6(s,3H),1.0(s,6H). Anal.Calcd for C9H160: C,77.09; H,11.50. Found: C,76.95; H,11.42.

3,5,5-Trimethylcyclohex-3-en-1-yl aceta=
te(la).3,5,5-Trimethylcyclohex-3-en-1-ol
(14.0 g, 9.9xlo mol) dissolved in gla=
cial acetic acid (30 ml) and acetic anhy=
dride (30 ml) was treated at reflux under
nitrogen. After 1 h the solution was co=
oled, added with diethyl ether (150 ml)
and washed with water (3x70 ml) and suc=
cessively with brine (3x30 ml). The or=
ganic phase was dried (Na₂SO₄) and the
solvent was distilled under reduced pres=
sure to give 17.10 g (94%) of la as a

colorless liquid: bp 87-89°C(9.0 mm_{Hg}); IR(neat) 1740 (s), 1670 (m)cm⁻¹; 1 H NMR δ 5.1(s,1H),4.0(m,1H),2.1-1.7(m,4H),1.6 (s,3H).Anal.Calcd for $^{11}_{18}^{6}_{2}$: C,72.49; H,9.96. Found: C,72.35; H,9.85.

3,5,5-Trimethylcyclohex-3-en-1-yl benzo= ate (lb).Benzoyl chloride (ll.1,7.9xlo-2 mol) in dry benzene (40 ml) was slowly added at 10-15°C with stirring to a so= lution of 3,5,5-trimethylcyclohex-3-en-1-ol (1) (10.0 g, 7.1×10^{-2} mol), pyridine $(28.4 \text{ g}, 36 \text{x} 10^{-2} \text{mol})$ and benzene (100 ml). The reaction mixture was refluxed for 2 h , and then stirred for an addi= tional hour at room temperature; 10% a= queous HCl was added until acidic pH was reached, and the organic phase was sepa= rated and washed with brine, dried (Na2SO4) and the solvent was distilled under reduced pressure. Distillation of residual oil yielded 11.03 g (63% yield) of 1b as a colorless liquid: bp 100-101° C(0.3 mm_{Hq}); IR(neat) 1720(s),1670(m) cm^{-1} ; ¹ H NMR δ 5.35 (m, 5H), 5.2(s, 1H), 2.4-1.6(m,4H),1.6(s,3H),1.0(s,6H). Anal.Cal cd for C₁₆H₂₀O₂: C,78.65; H,8.25. Found: C,78.72; H,8.37.

3,5,5-Trimethyl-1-(tetrhydrofuranyloxy) cyclohex-3-ene(lc). Amberlist H-15,H+ form (1 g), was added to a cooled solu= tion of compound 1 $(9.79 \text{ g,6.9x10}^{-2} \text{mol})$ and dihydrofuran (26.4x10⁻²mol) in CH₂Cl₂ (200 ml), after the mixture was stirred for 3 h at room temperature, the resin was taken away by decantation and the residue was washed with a 10% aqueous NaHCO2, with saturated brine and finally dried (Na2SO4). The solvent was distil= led at reduced pressure. Distillation of residual oil yielded 9.14 g (63% yield) of lc as colorless liquid: bp 58-60°C (0.3 mm_{Hg}); IR(neat) 1670(m) cm⁻¹; ¹H NMR δ 5.35(s,1H),5.10(s,1H),3.9(m,3H),2.2 -1.6(m, 8H), 1.6(s, 3H), 1.0(s, 6H). Anal.Cal= cd for C₁₃H₂₂O₂: C,74.24; H,10.54.Found: C,74.38; H,10.62.

1,2-Dicarbomethoxy-4-methylcyclohex-4ene(3). Few drops of37% HCl were added to the solution of compound 2 (15 g,9.0 $x10^{-2}$ mol) in MeOH (80 ml). The mixture was distilled at reduced pressure, die= thyl ether was added and the ethereal solution was washed with saturated aque= ous NaHCO, and brine, dried (Na,SO,) and concentrated at reduced pressure to yield an oil. Fractional distillation produced 13.7 g(69% yield) of 3 as colorless li= quid: bp 100-102°C(1.5 mm_{Hg}); IR(neat) 1735(s),1670(m) cm⁻¹; ¹H NMR **ô**5.4(m,1H), 3.7(s,6H),3.0(m,2H),2.35(m,4H),1.75(s, 3H). Anal.Calcd for C₁₁H₁₆O₄: C,62.25; H,7.60. Found: C,62.32; H,7.68.

Cycloaddition of Dichloroketene with Compound la.Preparation of 8,8-Dichloro-1,5, 5-trimethylbicyclo [4.2.0] octan-7-on-3-yl acetate (4) and 7,7-Dichloro-1,5,5-trime= thylbicyclo[4.2.0]octan-8-on-3-yl acetate (5). A 250 ml three necked flask equip= ped with a condenser, addition funnel, ma= gnetic stirrer and No inlet was flame dried while purged with N2. When cooled the flask was charged with activated zinc dust $(5.0 \text{ g}, 7.6 \times 10^{-2} \text{mol})$, compound <u>la</u> $(3.0 \text{ g}, 1.6 \times 10^{-2} \text{mol})$ and 60 ml of anhy= drous ether. The suspension was stirred under N₂ at reflux and a solution of tri= chloroacetyl chloride (6.64 g, 3.65xl0⁻² mol) and phosphorus oxychloride (5.61 g, 3.5×10^{-2} mol) in 30 ml of anhydrous ether was added dropwise over a 3 h period. When the addition of the solution complete, the mixture was refluxed with stirring for additional 10 h. The reaction mixture was cooled and then was filtered through a pad of Celite and the unreacted zinc washed with ether (2x50 ml). The ethereal solution was washed successively with water (2x25 ml), a cold saturated

NaHCO₃ solution (2x25 ml) and brine (2x 25 ml) and dried (Na₂SO₄). The solvent was removed at reduced pressure to leave 5.6 g of a mixture, which was chromato= graphed on a silica gel column (benzenehexane, 8:2) eluting at first unreacted ester <u>la</u> and successively compound <u>4</u> (1.67 g, 35% yield) and compound <u>5</u> (1.31 g, 28% yield).

 $\frac{8,8-\text{Dichloro-1,5,5,5-trimethylbicyclo} \boxed{4}}{2.0 \text{ loctan-7-on-3-y1 acetate}} (4) \text{ was an oil: IR(neat),1805(s),1735(s)cm}^{-1}; ^{1}\text{H NMR}} \\ \delta 5.1(\text{m,1H}),3.40(\text{bs,1H}),2.10(\text{s,3H}),1.68} \\ (\text{s,3H}),1.22(\text{s,3H}),1.12(\text{s,3H}),3.1-1.9(\text{m,4H}). Anal.Calcd for $C_{13}^{\text{H}}_{18}^{\text{O}}_{3}^{\text{Cl}}_{2}$: $C,53.65$; H,6.13. Found: $C,53.74$; H,5.97.}$

7,7-Dichloro-1,5,5-trimethylbicyclo [4.2. O]octan-8-on-3-y1 acetate (5) was an oil: IR(neat) 1805(s),1735(s)cm⁻¹; ¹H NMR 04.95(m,1H),3.28(bs,1H)2.12(s,3H),1.62(s,3H),1.28(s,3H),1.04(s,3H),3.1-1.95(m,4H). Anal.Calcd for $C_{13}^{H}_{18}O_{3}^{Cl}_{2}$: C,53.65;H,6.13; Found: C,53.69; H,6.18.

Dechlorination of Dichlorobicyclo 4.2.0 octane Derivatives 4 and 5.Preparation of 1,5,5-Trimethylbicyclo [4.2.0] octan-7-on-3-yl acetate (6) and 1,5,5-Trimethyl bicyclo 4.2.0 octan-8-on-3-yl acetate (7) The mixture of compound $\underline{4}$ and $\underline{5}$ (4 g,1.36 $x10^{-2}$ mol) and zinc dust(8.0 g,12.3x10⁻² mol) was added to saturated methanolic solution of NHACl (60 ml) in a 100 ml flask equipped with a condenser. The su= spension was refluxed for 4 h, cooled and filtered. Methanol was evaporated at re= duced pressure and the residue was dis= solved with diethyl ether. The ethereal solution was washed with water and brine and dried (Na,SO,). The solvent was removed at reduced pressure to leave an oil (2.82 g,92%) which was chromatographed on a silica gel column (benzene-ethyl acetate, 97:3), eluting at first compound $\underline{6}$ (1.78 g) and successively compound $\underline{7}$

(1.04 g).

1,5,5-Trimethylbicyclo [4.2.0] octan-7-on

3-yl acetate (6) was anoil: IR(neat) 1775
(s),1730(s)cm⁻¹; H NMR 05.12(m,1H),2.70

(s,3H),2.04(s,3H),1.54(s,3H),1.18(s,3H), 1.08(s,3H),2.0-1.5(m,4H).Anal.Calcd for

C₁₃H₂₀O₃: C,69.61; H,8.99. Found: C,70.03 H,9.07.

1,5,5-Trimethylbicyclo[4.2.0] octan-8-on 3-yl acetate (7) was an oil: IR(neat) 1775(s),1730(s) cm⁻¹; H NMR & 5.0(m,1H), 2.75-2.55(m,3H),2.00(s,3H),1.52(s,3H), 1.22(s,3H),1.04(s,3H),2.5-1.0(m,4H).

Anal.Calcd for C₁₃H₂₀O₃: C,69.61; H,8.99. Found: C,69.85; H,8.75.

Cycloaddition of Dichloroketene with Com= pound lb.Preparation of 1,5,5-Trimethyl= bicyclo 4.2.0 octan-7-on-3-yl benzoate (10), and 1,5,5-Trimethylbicyclo 4.2.0 oc= tan-8-on-3-yl benzoate (11). According to the previously reported procedure, a mix= ture of dichlorobicyclo 4.2.0 octanones 8 and 9 obtained as unstable liquid from activated zinc dust (6.0 g,9.1x10 mol), compound lb (4.64 g,1.9xl0⁻²mol) trichlo= roacetylchloride (8.36 g,4.6x10⁻²mol) and phosphorus oxychloride $(4.2 \text{ ml}, 4.6 \text{x} 10^{-2})$. The crude mixture was dechlorinated by treatmentwith zinc dust (6.0 g,9.1x10⁻² mol) in saturated methanolic solution of NH Cl. Usual work-up of reaction yielded 3.91 g of an oil which was chromatographed on a silica gel column (benzene-ethyl-ace= tate, 97:3) eluting at first unreacted ester 1b, compound 11 (1.20 g) and succes= sively compound 10 (0.80 g). 1,5,5-Trimethylbicyclo 4.2.0 octan-7-on-3

<u>yl benzoate</u> (<u>10</u>) was a wite solid: mp 76-78°C; IR(KBr) 1770(s),1705(m)cm⁻¹; H NMR & 8.15-8.00(m,2H),7.65-7.30(m,3H),5.42(m,1H),2.75(s,3H),2.35-1.70(m,4H),1.55(s,3H),1.20(s,3H),1.10(s,3H).Anal. Calcd for

C₁₈H₂₂O₃: C,75.49; H,7.74. Found: C,75.84;

H,7.81.

 $\frac{1,5,5-\text{Trimethylbicyclo}\left[4.2.0\right]\text{octan-8-on}}{3-\text{yl benzoate}} \quad \frac{(11)}{1780(s)}, 1175(m)\text{cm}^{-1}; ^{1}\text{H NMR} \\ \delta & 8.15-7.98 \\ (m,2H)), 7.65-7.30(m,3H)), 5.30(m,1H), 2.76 \\ -2.58(m,3H), 2.50-1.30(m,4H), 1.60(s,3H), \\ 1.30(s,3H), 1.12(s,3H). \\ \text{Anal.Calcd for C}_{18} \\ \text{H}_{22}\text{O}_{3}: C,75.49; H,7.74.Found: C,75.61; H,7.35.}$

Cycloaddition of Dichloroketene with Com= pound 3. Preparation of 3,4-Dicarbometho= xy-8,8-dichloro-1-methylbicyclo[4.2.0]oc= tan-7-one (12). Compound 3 (4.03g,1.9x 10⁻²mol) was treated with activated zinc dust (6.0,9.1x10⁻²mol),trichloroacety1= chloride (5.0 ml, 4.6×10^{-2} mol) and pho= sphorus oxvenloride $(4.2 \text{ ml.} 4.6 \text{x} 10^{-2} \text{mol})$ according to the previous procedure to yield 3,4-dicarbomethoxy-8,8-dichloro-1methylbicyclo 4.2.0 octan-7-one (12)(3.98 g,65% yield) as pale yellow crystals:mp $67-68^{\circ}$ C; IR(KBr) 1795(s), 1740(s) cm⁻¹; H NMR δ 3.70(s,3H),3.62(m,1H),3.38-3.18(m, 2H),2.68-1.90(m,4H),1.42(s,3H). Anal. Calcd for $C_{13}H_{16}Cl_2O_5$: C,48.31; H,5.56. Found: C,48.48; H,5.67.

Preparation of 3,4-Dicarbomethoxy-1-me=

thylbicyclo [4.2.0] octan-7-one (13). Com=

pound 12 (2.0 g,0.6x10⁻²mo1) was dechlo=

rinated by treatment with zinc dust (3.0 g,4.55x10⁻²mo1) in saturated methanol

solution of NH₄C1 (20 m1).After 4h at

reflux, usual work-up of reaction mixtu=

re furnished 1.42 g of compound 13 as an

oil that was purified by silica gel co=

lumn chromatography (Et₂0-hexane 1:1):

IR(neat)1775(s),1735(s) cm⁻¹; h NMR δ

3.76(s,6h),3.30-3.00(m,2h),2.70-1.55(m,

5H),1.4(s,3h). Anal.Calcd for C₁₃H₁₈O₅:

C, 61.40; H,7.14. Found: C,61.27; H,7.65.

Lactone of cis-2-Hydroxy-4-benzoyloxy-2, 6,6-trimethyl-1-acetic Acid (14). To a solution of compound 10 (0.40 g,0.14x10⁻² mol) in chloroform (10 ml) cooled at 0°C was added MCPBA (0.26 g,0.15xlo⁻²mol).

The reaction mixture was stirred for 24 h at 0°C. The product was extracted with ether and washed with 10% aqueous sodium sulfite and saturated sodium carbonate.

The ether layer was dried over MgSO₄ and the solvet was removed in vacuo. Compound 14 (0.340 g, 90% yield) is a solid: mp 102-104°C; IR(KBr): 1725(s),1715(s) cm⁻¹; 1H NMR 68.16-7.00(m,2H),7.60-7.20(m,3H), 5.30(m,1H),3.90(s,1H),2.35(s,2H),1.90-1.60(m,4H),1.4(s,3H),1.30(s,3H),1.18(s,3H).

Anal.Calcd for C₁₈H₂₂O₄: C,71.54; H,7.34. Found: C,71.74; H,7.42.

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